## 中钨在线(厦门)科技有限公司 CHINATUNGSTEN ONLINE (XIAMEN) MANU. & SALES CORP. 厦门市软件园二期望海路 25 号 3 楼 <u>www.chinatungsten.com</u> <u>sales@chinatungsten.com</u> 传真: 0592 512 9797 电话: 0592 512 9696/512 9595/ 13806045308

# Preparation Method of Self-Assembly W18O49 Nanostructure CN 103570070 A

## ABSTRACT

The invention provides a preparation method of a self-assembly W18049 nanostructure. The preparation method is characterized by comprising the following steps: weighing WC16 and ZnCl2; dissolving the WC16 and the ZnCl2 into a mixed solvent of ethanol and polyethylene glycol 400; transferring into a reaction kettle, and reacting under the condition of 160-220 DEG C for 18-24 hours; centrifugalizing; carrying out solid-liquid separation; and washing to obtain a solid, namely the self-assembly W18049 nanostructure. The preparation method provided by the invention can be used for preparing the self-assembly W18049 nanostructure by taking a common precursor as a raw material and using a simple experiment device; the prepared self-assembly W18049 nanostructure ranges from 700 micrometers to 1.4 micrometers on structural size and comprises nanometer bundles of 5-20 nanometers. The self-assembly W18049 nanostructure provided by the invention can be well dispersed in water and achieves very high absorption in a near-infrared region; the prepared self-assembly W18049 nanostructure is expected to be used for the fields of photoelectricity, nano-biology and the like.

## DESCRIPTION

Kind of self-assembled nanostructures preparation W18O49

## Technical Field

The present invention belongs to the field of nano material preparation, particularly, to a self-assembly method stomach 18049 nano structures.

## Background

The self-assembled nano structures are nano-materials for optical applications, optoelectronic applications - important prerequisite. Self-assembled nanostructures with respect to the single - nanostructure has a more excellent properties (J.Am.Chem.Soc.2011,133,15946). As tungsten oxide species important semiconductor material, in many areas has been widely used, it has been of great concern. For example, oxidation crane (W03 x) nano-materials have been used with electrochromic sensor (Adv.Funct.Mater.2008,18,1922), the gas window (Chem.Eur.J.2011,17,5145), photocatalysis (Small2008, 4,1813), photothermal of cancer (Adv.Mater.2013,25,2095). Currently treatment reported W03 x nanostructures mainly nanowires and nanorods, and self-assembled nanostructures W18O49 been reported.

The present invention is a common precursor in the synthesis of low-cost assembly W18O49 nanostructures. No simple equipment requirements. Product defects due to the presence of oxygen in the near-infrared region has a strong absorption.

## DISCLOSURE

The object of the present invention is to provide a low-cost - kinds of commonly used synthetic precursor W18O49 nanostructure self-assembly method.

To achieve the above objects, the present invention provides a method of self-assembled nano W18O49 structure, characterized in that the specific steps include: Weigh WCl6 and ZnCl2, which was dissolved ko mixing alcohol and polyethylene glycol 400 solvents, transferred to the reaction dad, in the 160-220 ° C reaction conditions 18-24h, centrifugation, solid-liquid separation, washing the resulting solid paint to give self-assembled nanostructures W18O49.

Preferably, the molar ratio of said WCl6 and ZnCl2 as I: 0.3-0.9.

Preferably, ko mixed solvent of the alcohol and polyethylene glycol 400 in a volume ratio ko and polyethylene glycol 400 is 1: 0.25.

Compared with the prior art, the beneficial effects of the present invention are:

1, the present invention is a common precursor for raw materials, with a simple experimental device can be prepared self-assembled nanostructures W18O49; preparation of self-assembled nanostructures W18O49 700-1.4iim size within the range from 5-20nm nano-bundles. Can be well dispersed in water the self-assembled structures in the near infrared region has a strong absorption. W18O49 self-assembled nanostructures could be used for the preparation of optical, nano-biotechnology and other fields.

2, the process used for the preparation process of the present invention is simple, cheap and readily available raw materials, high repetition rate.

**Brief Description** 

Figure 1 is a self-assembling stomach invention prepared 18049 nm low-fold structure, high magnification SEM images and high resolution transmission diagram;

FIG. 2 of the present invention prepared in the self-assembly of nano-XRD W18O49 image structure;

FIG. 3 of the present invention prepared in the self-assembly of nano W18O49 absorption spectrum structure; DETAILED DESCRIPTION

In order that the invention may be more fully understood, hereby to a preferred embodiment, described in detail below.

## Example 1

Weigh 0.1980g WCl6,0.0245g ZnCl2, dissolved in 40mL ko IOmL alcohol solvent mixture of polyethylene glycol 400, transferred to 60mL volume of reactor, under reaction conditions 180 ° C 24h, centrifuged, The supernatant was discarded, the resulting solid was washed, to give a blue product, i.e. W18O49 self-assembled nanostructures. Figure Ia, Ib description prepared by the W18O49 5-20nm nanowire bundle of assembled. Figure 1c shows that the self-assembled structure along the (010) direction of growth, and Figure 2, respectively. 3 for the structure of a UV - visible absorption spectrum, indicating that the self-assembled structures have a strong near-infrared absorption, could be used for light and heat treatment.

#### Example 2

Weigh 0.1980g WCl6,0.0245g ZnCl2, soluble in alcohol and 20mL 30mL ko mixed solvent of polyethylene glycol 400, transferred to 60mL volume of reactor, the reaction conditions at 180 ° C 24h. Centrifugation, the supernatant was discarded, the resulting solid was washed, to give a blue product, i.e. W18O49 self-assembled nanostructures.

#### Example 3

Weigh 0.3960g WCl6,0.0490g ZnCl2, dissolved in 40mL ko IOmL alcohol solvent mixture of polyethylene glycol 400, transferred to 60mL volume of reactor, the reaction conditions at 180 ° C 24h. Centrifugation, the supernatant was discarded, the resulting solid was washed, to give a blue product, i.e. W18O49 self-assembled nanostructures.

#### Example 4

Weigh 0.1980g WCl6,0.0245g ZnCl2, soluble in alcohol and 20mL 30mL ko mixed solvent of polyethylene glycol 400, transferred to 60mL volume of reactor, the reaction conditions at 180 ° C 12h. Centrifugation, the supernatant was discarded, the resulting solid was washed, to give a blue product, i.e. W18O49 self-assembled nanostructures.

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